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Solution: Rearrange $Q = mc\Delta t$ to solve for T .
 $Q = mc(T_2 - T_1)$
 $8.7 \times 10^2 \text{ J} = (0.05 \text{ kg})(1.3 \times 10^2 \text{ J/kg} \cdot ^\circ\text{C}) (T_2 - 13.4^\circ\text{C})$
 $8.7 \times 10^2 \text{ J} = 6.5 \text{ J/}^\circ\text{C} (T_2 - 13.4^\circ\text{C})$
 $134^\circ\text{C} + T_2 = 156^\circ\text{C}$
 $T_2 = 160^\circ\text{C}$
Statement: The final temperature of the lead was 160°C .

Chapter 6 Review, Understanding pages 310-315

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